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GERMACRANOLIDES FROM ERLANGEA CORDIFOLIA*

ABDOLHOSSEIN RUSTALYANT, LILLY NAZARIANST and FERDINAND BOHLMANNS

†Department of Chemistry, National University of Iran, Teheran, Iran; ‡Institut für Organische Chemie, Technische Universität Berlin, Strasse des 17. Juni 135, D-1000 Berlin 12, West Germany

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INTRODUCTION

So far only 3 members of the large African genus Erlangea (Compositae, tribe Vernonieae) have been investigated. In addition to simple sesquiterpenes, one species contains two guaianolides [1], while in a second one germacranolide of the glaucolide type was found [2]. The third species only contains special 5-methylcoumarins [3] also present in Bothriocline [4] and Ethulia [5], both being African genera belonging to the Vernonieae. A germacranolide is reported in a patent application from E. cordifolia [6], however, no stereochemistry was given. We now report the isolation and structure elucidation of two highly oxygenated germacranolides from the same plant, which are named cordifene and 4,15-epoxy-4,15-dihydrocordifene.

RESULTS AND DISCUSSION

The polar fractions of the aerial parts of $E.\ cordifolia$ afforded, as the main constituent, a crystal-line compound with the molecular formula $C_{20}H_{24}O_{7}$. The ¹H NMR spectrum (Table 1) was very similar to the spectrum reported by Mugo [6]. The 270 MHz ¹H NMR spectrum displayed typical signals for a methylene lactone (s(br), 6.31 and 5.67) and those for an angelic acid ester (qq, 6.08; dq, 1.93 and dq, 1.79). Extensive double resonance experiments lead to the sequences A and B.

Irradiation of the signal for H_h (dd, 3.00) collapsed the doublet at 2.60 and the triplet doublet at 3.97 to singlets. Irradiation at 3.97 sharpened the signals of the methylene protons at 5.53 and 5.39. The coupling $J_{\rm H_2H_2}$ was 7.5 Hz indicating a cis configuration of 1- and 2-H as in mikanolide [7, 8]. As the signal at 3.60 showed no further couplings and as there was a downfield methyl singlet at 1.58, the position of this group seems to be established. On irradiation of H_b (d(br), 4.38), a doublet at 4.12 H_o changes to a singlet, while irradiation of H_i sharpened the broadened singlets at 6.31 and 5.67 (13-H), and the ddd 5.11 became a dd. This clearly showed that the signal at 2.92 must be assigned to 7-H. Further irradiation of H_k (ddd, 5.11) collapsed the two doubledoublets at 2.57 and 1.34 to doublets. These results are only in agreement with part B. However, the relative positions of the oxygen functions remained to be elucidated. On acetylation a monoacetate was obtained. In the ¹H NMR spectrum (Table 1) the signal

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Table 1. ¹H NMR data of compounds **1, 2** and **5** (270 MHz, TMS as internal standard)

	1	Δ^*	2	5
1-H	2.60 d	0.44	2.60 d	2.63 d
2-H	3.00 dd	0.37	3.05 dd	3.06 dd
3-H	3.97 dt	0.29	$3.80 \ d(br)$	$3.83 \ d(br)$
5α-H	4.12 d	0.25	5.39 d	$3.23 \ d(br)$
6β-Н	4.38 d	0.31	4.46 d	4.53 d
7α-H	$2.92 \ d(br)$	0.21	$2.95 \ d(br)$	$2.93 \ d(br)$
8β-H	5.11 ddd	0.22	5.10 ddd	5.14 ddd
9x-H	1.34 dd	0.24	1.34 dd	1.43 dd
9β-H	2.57 dd	0.18	2.59 dd	2.63 dd
13-H	$5.67 \ s(br)$	0.10	5.70 s(br)	$5.67 \ s(br)$
13'-H	6.31 s(br)	0.14	$6.37 \ s(br)$	$6.30 \ s(br)$
15-H	$5.29 \ s(br)$	0.12	5.44 s(br)	3.29 d
15'-H	$5.53 \ s(br)$	0.24	$5.63 \ s(br)$	$2.73 \ d(br)$
14-H	1.58 s	0.16	1.60 s	1.64 s
18-H	6.08 qq	0.04	6.09 qq	6.11 qq
19-H	1.93 dq	0.04	1.93 dq	1.94 dq
20-H	1.79 dq	0.04	1.78 dq	1.80 dq
—ОН	3.15 s(br)	0.22		3.14 s(br)
OAc	-	-	2.16 s	

^{*}Δ values after addition of Eu(fod)₃.

J (Hz): 1,2=7.5; 2.3=4.5; $3.15\sim1$; 5.6=9.5; 7.8=10; $7.13\sim1$; $8.9\alpha=12$; $8.9\beta=4$; $9\alpha.9\beta=12.5$; 18.19=7.5; 18.20=19.20=1.5; bei **5**: 15.15'=5; $15'.6\sim1$.

for H_a was shifted downfield. This indicated that we were dealing with a germacranolide. However, the stereochemistry and the relative position of the oxygen function still was not clear. It was obvious from the NMR data that the couplings $J_{6,7}$ and $J_{7,8}$ must be very small. Inspection of Dreiding-models showed that we were most probably dealing with a 6.12-transfused lactone in a conformation with C-14 and C-15 above the plane. The other observed couplings would be in agreement with 1. However, the second possibility would be 3. Considering the position of the signal for proton at the OH-bearing carbon, an allylic position as in 1 seemed to be more probable. Furthermore, the observed shifts on acetylation indicated this position as there was a pronounced shift of the signals of the methylene protons (15-H). All efforts to transform the alcohol to the corresponding ketone were unsuccessful. MnO₂ as well as pyridine chlorochromate gave no identifiable products. The stereochemistry at C-5, C-6 and C-8 followed from the observed couplings, while that at C-10 remained uncertain. We have assumed this configuration by analogy to similar lactones. All data therefore were in good agreement with the proposed structure 1. Also the ¹³C NMR data (Table 2) supported this assumption. The assignments of most of the signals were easy. However, some may be interchangeable. The unusual positions of the signal of C-8 is probably due to the shielding effect of the C-10 methyl, which support the configuration at C-10.

A second slightly more polar lactone, molecular formula $C_{20}H_{24}O_8$, showed a very similar ¹H NMR spectrum to 1. The main difference was the absence of the vinylic proton signals for 15-H. They were replaced by a doublet at 3.29 and a broadened doublet at 2.73.

The last signal showed a W-coupling with 5-H. The positions of these signals and the coupling constant indicated the presence of an epoxide of 1. Though the configuration of this new group could not be established, the NMR data were in good agreement with the α -epoxide 5. The observed shift of the signal for 6-H if compared with that in 1 again showed that the general structure 1 should be favoured, since it could not be explained by an epoxide of 3. Therefore, most probably the second lactone was 4,15-epoxy-4,15-dihydrocoidifene (5).

The isolation of 1 and 5 from an Erlangea species again shows that highly oxygenated sesquiterpene lactones are characteristic for this genus. However, these compounds are also present in the large genus Vernonia. The chemotaxonomic situation in this tribe is

Table 2. ¹³C NMR chemical shifts of compound 1 (CDCl₃)

F					
C-1	52.4 d*	C-11	134.0 s		
C-2	54.2 d*	C-12	167.8 s		
C-3	62.4 d	C-13	127.1 t		
C-4	139.5 s	C-14	118.6 t		
C-5	80.6 d	C-15	17.8 q		
C-6	77.8 d	C-16	166.0 s		
C-7	47.9 d	C-17	$127.0 \ s$		
C-8	67.9 d	C-18	140.0 d		
C-9	45.4 t	C-19	15.8 q		
C-10	55.2 s	C-20	20.1 q		

^{*}Possibly interchangeable.

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still not very clear, but as pointed out by Jones [9] the systematic situation is also confused. Perhaps further chemical investigations may be helpful.

EXPERIMENTAL.

IR: CHCl₃; ¹H NMR: 270 MHz, TMS as int. standard; MS: 70 eV; optical rotation: CHCl₃. The air-dried plant material (29 g aerial parts) was extracted with Et₂O at room temp. Saturated hydrocarbons were removed by treating with MeOH. The resulting extract was first separated by CC (Si gel, act. grade II). With Et₂O 1 and 5 were eluted, which could be separated by repeated TLC (Si gel GF 254) (Et₂Opetrol, 7:3) yielding ca 30 mg 1 and 15 mg 5.

Cordifene (1). Colourless crystals from Et₂O-petrol, mp 183°. IR $\lambda_{max}^{CHCl_3}$ cm $^{-1}$: 3580 (OH), 1780 (γ-lactone), 1720 (C=CCO₂R); MS m/e (rel. int.): 376 (M $^+$ 0.6%) C₂₀H₂₄O₄); 361 (- Me, 0.3); 358 (- H₂O, 1); 276 (-C₄H₇CO₂H, 2); 83 (C₄H₇CO $^+$, 100); 55 (83 - CO, 70).

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{-127.1} \frac{578}{-133.9} \frac{546}{-154.8} \frac{436}{-274.2} (c = 0.31).$$

5 mg **1** in 0.5 ml Ac₂O was heated for 30 min at 70°. After evapn the residue was purified by TLC (Et₂O-petrol, 7:3) yielding 3 mg of **2**, colourless oil, IR $\nu_{\rm cc}^{\rm CHCl_3}$ cm⁻¹: 1800 (lactone), 1760, 1230 (OAc), 1730 (C—CCO₂R), MS (Cl, isobutane m/e (rel. int.): M+1-; 391 (-CO, 100); 377 (-ketene, 49); 289 (-CHO, C₄H₇CO₂H, 30); 229 (289-HOAc, 17). Oxidation of **1** with MnO₂ or pyridine chloro-

chromate led to degradation of the lactone. Definite products could not be isolated.

4,15-Epoxy-4,15-dihydrocordifene (5). Colourless gum, IR $\nu_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$: 3580 (OH), 1780 (lactone), 1720 (C=CCO $_2$ R); 1 H NMR see Table 1.

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